

1 IN THE CLAIMS

2 CLAIMS

3 1. Cancelled.

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5 (CURRENTLY AMENDED) 72. A process for treating wood having wood cellulose having a
6 plurality of hydroxyl groups comprising the steps of:

7 providing a solution consisting essentially of a water compatible organic solvent and a
8 dissolved solute having a functional group comprising an atom selected from the group
9 consisting of trivalent, tetravalent and pentavalent atoms, wherein said atom is bonded to a
10 halogen atom or a functional group selected from the group consisting of a hydroxyl group,
11 alkoxy group, phenoxy group, benzyloxy group and an aryloxy group having a polycyclic
12 aromatic ring,

13 and wherein the organic solvent is further defined as capable of allowing the dissolved
14 solutes to be drawn from the solute into the wood;

15 adding a catalyst reacting with water in the wood to produce an acid;

16 applying said solution to the wood;

17 drawing the catalyst and solute into the wood;

18 [hydrolyzing] reacting the catalyst with [the water in] the wood to produce a solvated acid;
19 covalently reacting said functional groups using the solvated acid with the cellulose.

20 (PREVIOUSLY PRESENTED) 73. The process according to claim 76 wherein the solutes are not
21 polymerized prior to drawing the catalyst into the wood.

22 (CURRENTLY AMENDED) 74. The invention of claim 72 wherein heat is generated by the acid
23 solutes reacting with the cellulose.

1 (PREVIOUSLY PRESENTED) 75. The process according to claim 72 wherein the process of
2 drawing further comprises water in the wood pulling the solutes into from the solvent into the wood
3 and reacting the cellulose with the hydrolyzed acid and solute.

4 (PREVIOUSLY PRESENTED) 76. The process of claim 72 wherein the solute is comprised of
5 monomers prior to application of the solute to said wood.

6 (PREVIOUSLY PRESENTED) 77. The process of claim 72 wherein the catalyst is less than 10%
7 by mass of the entire solution.

8 (PREVIOUSLY PRESENTED) 78. The process of claim 72 wherein the step of covalently
9 reacting further comprises the step of catalytically bonding the functional group tetravalent atom
10 across an oxygen of the cellulose.

11 79. CANCELLED.

12 (PREVIOUSLY PRESENTED) 80. A process for treating wood having wood cellulose having a
13 plurality of hydroxyl groups comprising the steps of:

14 providing a solution consisting essentially of a water compatible organic solvent and a
15 solute having a functional group comprising an atom selected from the group consisting of
16 trivalent, tetravalent and pentavalent atoms, wherein said atom is bonded to a halogen atom or a
17 functional group selected from the group consisting of a hydroxyl group, alkoxy group, phenoxy
18 group, benzyloxy group and an aryloxy group having a polycyclic aromatic ring, and wherein the
19 solvent is further defined as allowing the solutes to be drawn from the solute into the wood;

20 adding a catalyst reacting with water in the wood to produce a base;

21 applying said solution to the wood;

22 drawing the catalyst and solute into the wood;

23 reacting the catalyst with the water in the wood to produce a hydrolyzed base;

1 covalently reacting said functional groups with the hydrolyzed base and the cellulose.

2 (PREVIOUSLY PRESENTED) 81. The process of claim 72 wherein the acid is a strong acid.

3 (PREVIOUSLY PRESENTED) 82. The process according to claim 80 wherein the step of reacting
4 further comprises reacting without adding heat or pressure to initiate the reaction prior to drawing
5 the catalyst into the wood.

6 (PREVIOUSLY PRESENTED) 83. The process of claim 77 wherein the catalyst is in the range
7 of 0.1-10% of the solution.

8 (PREVIOUSLY PRESENTED) 84. The process of claim 83 wherein the catalyst is in the range
9 from 0.1 to 4.9% of the solution.

10 (PREVIOUSLY PRESENTED) 85. The process of claim 81 wherein the acid is selected from
11 the group consisting of acids from alkyl-halide monomers with trivalent, tetravalent and pentavalent
12 atoms.

13 (PREVIOUSLY PRESENTED) 86. The process of claim 85 wherein the acid is comprised of
14 silicon and a halogen.

15 (PREVIOUSLY PRESENTED) 87. The process of claim 72 wherein the solute comprises a non-
16 catalyst producing molecule which reacts to covalently bond with wood cellulose in the presence of
17 acid from the catalyst.

18 (PREVIOUSLY PRESENTED) 88. The process of claim 87 wherein a molecule which does not
19 produce an acid in the presence of water in wood cellulose reacts to produce heat with wood
20 cellulose in the presence of the molecule producing an acid in the presence of water in wood
21 cellulose.

22 (PREVIOUSLY PRESENTED) 89. The process of claim 88 wherein the non-acid producing
23 molecule is from the group consisting of alkyl and hydroxyl or alkoxy bonded trivalent, pentavalent

1 and tetravalent atoms and combines thereof.

2 (PREVIOUSLY PRESENTED) 90. The process of claim 85 wherein the catalyst is from the group
3 consisting of hydrochloric, meta-phosphoric acid, poly-phosphoric acid, Phosphoric acid, and
4 combinations thereof.

5 (PREVIOUSLY PRESENTED) 91. The process of claim 85 wherein the acid is in the range of 0.01-
6 10% *in situ*.

7 (PREVIOUSLY PRESENTED) 92. The process of claim 72 wherein the process further comprises
8 avoiding water based formation of oligomers of the functional groups prior to applying said solution
9 to said wood.

10 (PREVIOUSLY PRESENTED) 93. The process of claim 72 further comprising the step of:

11 adding at least one non-reactive additive to the wood cellulose that enhances a desired
12 property selected from the group consisting of:

13 fire resistance,

14 insect resistance,

15 moisture resistance

16 color,

17 adhesion, and

18 insulation, and

19 combinations thereof.

20 (PREVIOUSLY PRESENTED) 94. The process of claim 93 wherein the step of adding at least one
21 non reactive additive further comprises adding the additive to the solution.

22 (PREVIOUSLY PRESENTED) 95. The process of claim 93 wherein the step of adding the at least
23 one non-reactive additive occurs before reacting the functional groups to bond with the wood

1 cellulose.

2 (PREVIOUSLY PRESENTED) 96. The process of claim 93 wherein the additive is selected from
3 the group consisting of:

4 diatomaceous earth,
5 sodium silicates,
6 boron or silicon salts,
7 boric acid,
8 trimethyl (trialkyl) borate,
9 Boron Halides (BF₃, BCl₃, etc.),
10 Boric Anhydride (boron oxide),
11 phosphorous compounds,
12 copper compounds,
13 metal alkoxide,
14 meta-phosphoric acid;
15 a hydrophobic reagents,
16 phosphoric acid, and
17 metaphosphoric acid,
18 and combinations thereof.

19 97. Cancelled.

20 (PREVIOUSLY PRESENTED) 98. The process according to claim 72, wherein the wood cellulose
21 has an original weight and wherein the duration of treatment attains a weight of a compound which
22 is covalently bonded to the wood cellulose in a range of 0.1 to 10 weight percent of the original
23 weight of the wood cellulose.

1 99. Cancelled.

2 (PREVIOUSLY PRESENTED) 100. The process of claim 81 further comprising the step of
3 exposing the acids introduced into the wood to an acid neutralizing agent subsequent to the
4 treatment.

5 (PREVIOUSLY PRESENTED) 101. The process of claim 81 further comprising the step of
6 introducing an acid neutralizing agent into the wood prior to the exposure of the wood cellulose to
7 the acid.

8 102. CANCELLED

9 103. CANCELLED

10 (PREVIOUSLY PRESENTED) 104. A process according to claim 72 wherein the wood
11 cellulose is not dry and wherein the step of drawing further comprises solvating the functional
12 groups by the water in the wood prior to being covalently bonded to the hydroxyl groups of said
13 wood cellulose.

14 (PREVIOUSLY PRESENTED) 105. The process according to claim 72 further comprising
15 the step of adding water to the wood cellulose prior to applying the solution to the wood cellulose.

16 (PREVIOUSLY PRESENTED) 106. A process for treating wood cellulose containing water,
17 said cellulose having a plurality of hydroxyl groups comprising the steps of:

18 providing a solution comprised of a water compatible organic solvent and a solute having
19 a plurality of unreacted solutes comprising an atom selected from the group consisting of
20 trivalent, tetravalent and pentavalent atoms, wherein said atom is bonded to a halogen atom or a
21 functional group selected from the group consisting of a hydroxyl group, alkoxy group, phenoxy
22 group, benzyloxy group and an aryloxy group having a polycyclic aromatic ring, applying said
23 solution to the wood cellulose; and simultaneously pulling said solution into the wood using the

1 water within the wood and reacting said solute to form covalent bonds, and forming a matrix
2 structure comprising reacted monomers and wood cellulose.

3 (PREVIOUSLY PRESENTED) 107. The process of claim 106 further comprising the step of:
4 adding at least one non-reactive additive that enhances a desired property selected from
5 the group consisting of:

6 fire resistance,
7 insect resistance,
8 moisture resistance
9 color,
10 adhesion, and
11 insulation, and
12 combinations thereof.

13 (PREVIOUSLY PRESENTED) 108. The process of claim 107 wherein the step of adding
14 the at least one non-reactive additive occurs before covalently bonding the compound to the
15 wood cellulose.

16 109. CANCELLED.

17 (PREVIOUSLY PRESENTED) 110. The process according to claim 106, further
18 comprising a step of exposing the wood to ultra-sound sonification while applying said solution.

19 (CURRENTLY AMENDED) 111. A process for treating wood cellulose, in wood having water
20 in the wood having a plurality of hydroxyl groups comprising the steps of:

21 providing a solution comprised of a water compatible organic solvent; an acid; an atom
22 selected from the group consisting of trivalent, tetravalent and pentavalent atoms, wherein said atom
23 is bonded to a halogen atom or a functional group selected from the group consisting of a hydroxyl

1 group, alkoxy group, phenoxy group, benzyloxy group and an aryloxy group having a polycyclic
2 aromatic ring, applying said solution to the wood cellulose; solvating the acid with the water in the
3 wood to produce a solvated acid and reacting the atom with the hydrolyzed acid and the wood
4 cellulose to produce heat and a silicone cellulose bond.

5 (PREVIOUSLY PRESENTED) 112. The process of claim 111 wherein the step of reacting
6 further comprises the step of pulling the atoms from the solvent into the wood.

7 113. Cancelled.

8 114. Cancelled.

9 115. Cancelled.

10 116. Cancelled.

11 117. Cancelled.

12 118. Cancelled.

13 (PREVIOUSLY PRESENTED) 119. The process of claim 81 wherein the acid is produced by a
14 molecule producing an acid in the presence of water in wood.

15 (PREVIOUSLY PRESENTED) 120. The process of claim 119 wherein the acid is in the
16 range of 0.1-10% of the solution.

17 (PREVIOUSLY PRESENTED) 121. The process of claim 119 wherein the acid is in the
18 range from 0.1 to 4.9% of the solution.

19 (CURRENTLY AMENDED) 122. The process of claim 119 wherein the molecule is a molecule
20 comprised of a metal and a halogen.

21 (PREVIOUSLY PRESENTED) 123. The process of claim 120 wherein the functional group
22 comprises a molecule being drawn from the into the wood from the organic solvent and therein
23 producing an acid in the presence of water in wood cellulose and wherein the acid and solute reacts

1 producing heat on application to wood at standard atmospheric temperature and pressure.

2 (CURRENTLY AMENDED) 124. A process for treating wood cellulose having a plurality of
3 hydroxyl groups comprising the steps of: providing a solution comprised of a non-water-based
4 hydrophilic organic solvent; a molecule which is drawn from the solution into the wood and
5 produces an acid in the presence of water in the wood cellulose diffused as a chemical from the
6 solution and bonding with wood in conjunction with water in the wood and generating in the
7 bonding a catalyst; said solution further comprising a molecule drawn from the wood into the wood
8 cellulose and not producing an acid in the presence of water in wood cellulose diffused as a
9 chemical from the solution and bonding with wood cellulose in the presence of the catalyst generated
10 by the molecule producing an acid in the presence of water in wood cellulose.

11 (PREVIOUSLY PRESENTED) 125. The process of claim 124 wherein the catalyst is an acid
12 produced by the molecule which produces an acid in the presence of water in the wood cellulose
13 is defined as a molecule producing an acid causing a spontaneous reaction of the molecule producing
14 an acid in the presence of water in wood cellulose.

15 (PREVIOUSLY PRESENTED) 126. The process of claim 125 wherein the acid or a
16 molecule which produces an acid in the presence of wood cellulose is in the range of 0.1-10% of the
17 solution.

18 (PREVIOUSLY PRESENTED) 127. The process of claim 125 wherein the acid or a molecule which
19 produces an acid in the presence of wood cellulose is in the range from 0.1-4.9% of the solution.

20 (PREVIOUSLY PRESENTED) 128. The process of claim 112 wherein the acid is selected from
21 the group consisting of acids from alkyl-silicon halides, acids from alkyl-hilade monomers with
22 trivalent, tetravalent and pentavalent atoms, hydrochloric, meta-phosphoric acid, poly-phosphoric
23 acid, and Phosphoric acid and combinations thereof, wherein the acid is in the range of 0.01-10%

1 *in situ.*

2 (PREVIOUSLY PRESENTED) 129. The process of claim 128 wherein a molecule which produces

3 an acid in the presence of water in wood cellulose is a molecule comprised of silicone and a halogen.

4 (PREVIOUSLY PRESENTED) 130. The process of claim 129 wherein a molecule which does not

5 produce an acid in the presence of water in wood cellulose reacts exothermically and spontaneously

6 with wood in the presence of a molecule which does produce an acid in the presence of water in the

7 wood cellulose.

8 (PREVIOUSLY PRESENTED) 131. The process of claim 130 wherein the molecule which does

9 not produce acid in the presence of water in the wood cellulose would include hydroxyl and

10 alkoxyl bonded tetravalent atoms.

1 OBJECTIONS TO THE SPECIFICATION:

2 6. The examiner alleges in paragraph 6 that the word “sodium silicate” is not supported in
3 the specification. On page 12 lines 17-19, the following language appears:

4 (f) Due to the molecular reaction of “WPTC” and the wood’s natural liquids, the
5 wood expels liquid while absorbing sodium silicate and borax. The treated wood
6 weighs approximately as much after treatment as before;”

7 On page 13 lines 13-14, the following language appears:

8 “WPTC may introduce Borax and Sodium Silicate into the wood molecules thereby
9 providing significant water, fire, rot and insect protection.”

10 On page 27 lines 20-22 and page 28 lines 1-2, the following language appears:

11 “Diatomaceous earth, sodium silicates, or other boron or silicon salts may be used as a
12 source of donor atoms. These may be mixed to provide intermediaries in solution which would,
13 working together, carry out the desired end product in the wood. Examples of products having
14 these qualities include boric acid, trimethyl (trialkyl) borate, Boron Halides (BF₃, BC₁₃, etc.), and
15 Boric Anhydride (boron oxide).”

16 On page 29 lines 8-12, the following language appears:

17 “It is believed, but uncertain that borax and sodium silicate can be trapped inside a
18 polymer shield formed by the reaction.

19 The current “WPTC” formula incorporating the carbon-silicon-halogen reagent, a boron
20 donor, borax, sodium silicate, metal or metalloid catalysts or enhancers with THF or it’s
21 equivalents as the solvent.”

22 9. The examiner alleges in paragraph 9 that there is not “a description of the solute being
23 monomeric before application to the wood” is not supported in the specification. On page 30

1 lines 1-7 of the '165 patent, it reads as follows:

2 "While no chemical process necessarily results in a single outcome. Figure 6-b shows an
3 approximation of the most likely end structure for cellulose without a complete replacement of
4 hydroxyl atoms in the chain when treated with a mixture of silicon and boron under the process
5 steps taught hereunder.

6 Figure 6a shows a less likely structural outcome which is improbable and is given more
7 for purposes of disclosing all manner of potentially allowable structures as opposed to the
8 structure considered a likely end product."

9 11. The examiner alleges in paragraph 11 that the original specification "does not disclose
10 that the functional groups are solvated by water" is not supported in the specification. On page
11 37, lines 1-7 of patent '165, it reads:

12 "A plurality of Applicant's reactive molecules may enter to the wood cellulose from a
13 solution as shown in Figure 17A. Here the solution is an alcohol 72 solvated solution, although
14 there may be trace amounts of water 71 and other organic solvents 70. A pro-catalyst 27
15 (MeSiCl₃ here) and a silicone donor 73 (MeSi(OCH₃)₃ here) are used to prevent the pro-catalyst
16 27 from adding too much acidity to the wood. The use of hydrophilic organic solvents and
17 monomers allows the reaction to begin and proceed by simple diffusion of the solvents and
18 reactants into the wood."

19 12. The examiner alleges in paragraph 12 that the specification as originally filed does not
20 disclose "cyclic interlocking molecules having as a part of the cyclic structure at least two
21 carbons within the cellulose and at least two of the atoms from the functional groups consisting
22 of trivalent, tetravalent, and pentavalent atoms". Figure 11 shows the described compounds.

23 14. The examiner alleges in paragraph 14 that new matter was added to the claims in the

1 11/18/2003 amendment but that was not in the application as originally filed. This has been
2 corrected in the claims.